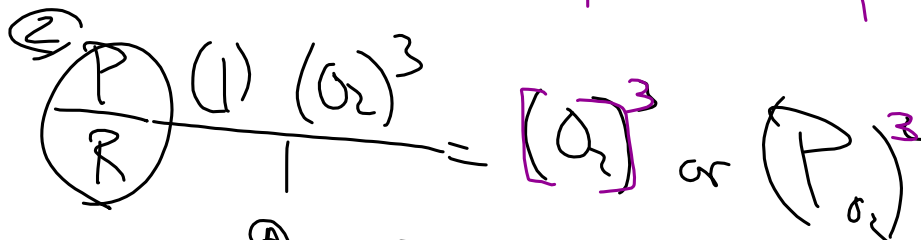
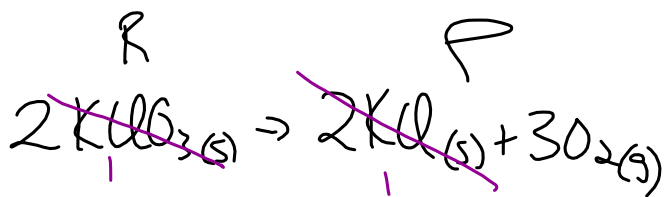


$$Q_k = \frac{[NH_3]^2}{[N_2][H_2]^3}$$



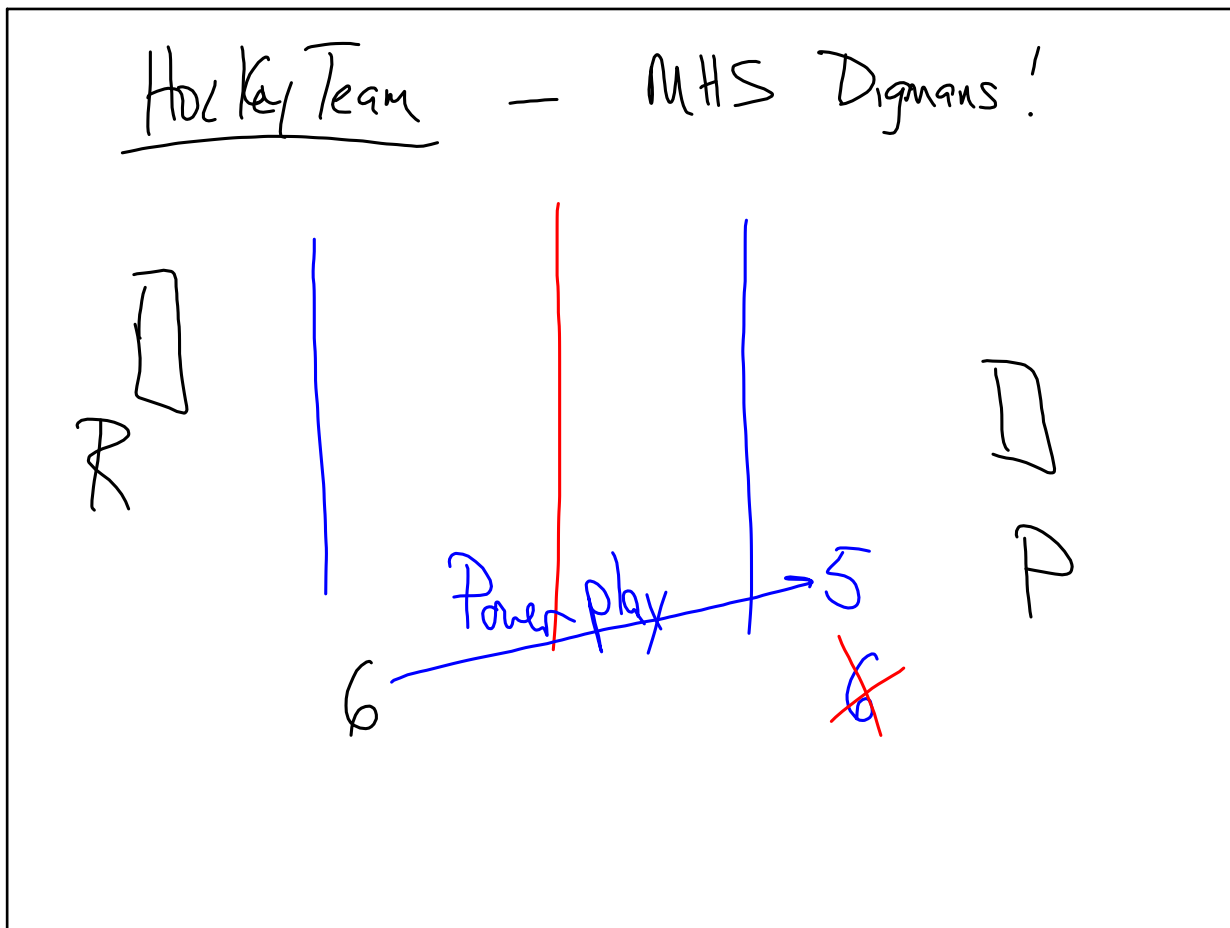
$$K = \frac{[H^+]}{[OH^-]}$$

Apr 6-9:21 AM

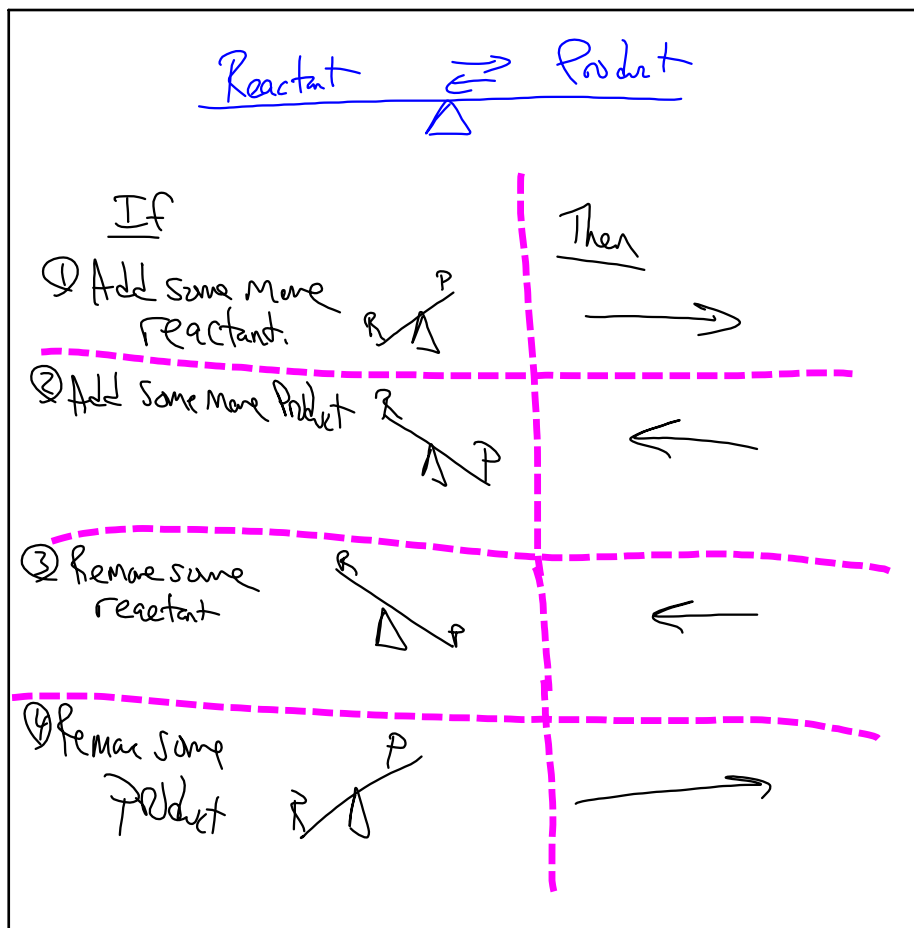
$$K = \frac{[CO_2]^2}{[CO]^2 [O_2]}$$

$$\frac{(Li^+)^2 [CO_3^{2-}]}{1}$$

Apr 6-9:49 AM



Apr 6-9:50 AM

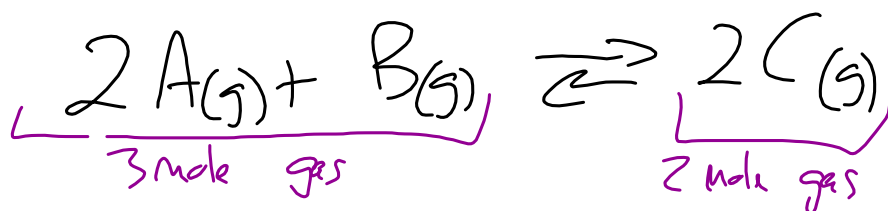



Apr 6-9:54 AM

Lechatlier NO STRESS!

IF you stress one side of
a rxn, The rxn moes to relieve
that stress and re achieve equilibrium

Apr 6-9:59 AM



IF 

Then 

Add Pressure
on both sides

Apr 6-10:01 AM



Add T 

decrease T 

Apr 6-10:02 AM