

BP elevation

$$\Delta T = (K_b \times m) i$$

H₂O $K_b = 0.52 \frac{^\circ\text{C}}{m}$

1m soln
2m soln

25ml H₂O

BaCl_2
 CaCl_2
 $\text{C}_6\text{H}_{12}\text{O}_6$
 $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$
 NaCl (1.45g / 1m) } 2.9g
 2 TOTAL

Jan 12-7:43 AM

1m salt soln (NaCl)

1m NaCl	58g NaCl	0.025 kg H ₂ O	= 1.45g NaCl
1 kg H ₂ O	1m NaCl		

Jan 12-7:55 AM