

PS 15 (K<sub>c</sub> = 9.7)

①  $\text{NH}_3(g) + \text{H}_2\text{S}(g) \rightleftharpoons \text{NH}_4\text{HS}(s)$

I	2M	2M	0
A	-x	-x	+x
E	2-x	2-x	x

0.3      0.3

1.7 M      0.34 mol/L

*mols?*

$$K_c = \frac{1}{[\text{NH}_3][\text{H}_2\text{S}]}$$

$$\frac{9.7}{1} = \frac{1}{(2-x)(2-x)}$$

$$\frac{9.7}{1} = \frac{1}{x^2 - 4x + 4}$$

$$9.7x^2 - 38.8x + 38.8 = 1$$

$$9.7x^2 - 38.8x + 37.8 = 0$$

$x_1 = 2.3$

$x_2 = 1.7$

Feb 13-7:36 AM

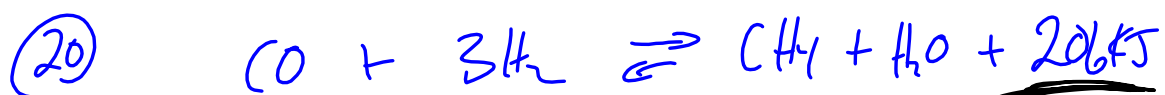
① E

$K_c = \frac{[P]}{[R]}$

$R + \text{heat} \rightarrow P$

As  $T \uparrow$ , favor more P,  
 $K_c \uparrow$


Feb 13-8:04 AM



4 moles (g)

2 mole (g)

Want HIGH P

Decr. T 

Feb 13-8:10 AM

(22) 
$$K_p = K_c (RT)^{\Delta n}$$

$$= 1.58 \times 10^{-8} \left[ 0.08206 (298) \right]^3$$

Feb 13-8:17 AM

(24)  $\text{NH}_4\text{HS (s)} \rightleftharpoons \text{NH}_3\text{(g)} + \text{H}_2\text{S(g)}$   $K_c = 1.6 \times 10^{-4}$

I	?	$0.8 \text{ M}$	$0$
Δ	$-x$	$+x$	$+x$
E	$x$	$0.8+x$	$x$

$K_c = \frac{[\text{NH}_3][\text{H}_2\text{S}]}{1}$

$(0.8+x)(x) = 1.6 \times 10^{-4}$

$x^2 + 0.8x - 1.6 \times 10^{-4} = 0$

$x = 2 \times 10^{-4}$

Feb 13-8:19 AM

SP Ex #1297

how much is left / have now!

(2)  $\ln A_t = -kt + \ln A_0$

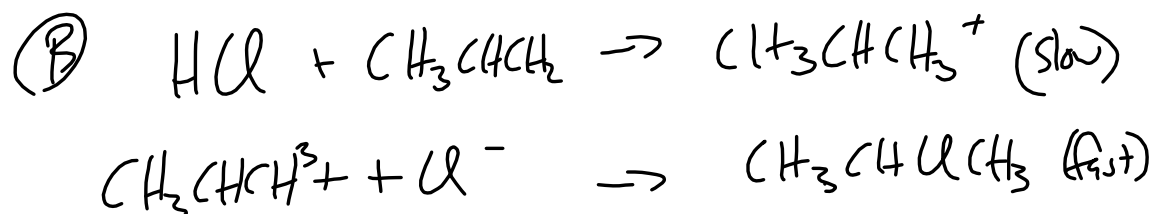
$\ln 0.15 = -k(79) + \ln A_0$

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(3)  $\frac{1}{A_t} = kt + \frac{1}{A_0}$

$\frac{1}{0.125} = (0.17)t + \frac{1}{0.25}$

Feb 13-8:35 AM



$$\textcircled{8} \quad \text{Rate} = k [\text{M}]^1 [\text{Z}]^1$$

$$1 \rightarrow 2 \quad 2' \quad = 2 \quad \left. \begin{array}{l} 2 \rightarrow 3 \\ 2' = 2 \end{array} \right\}$$

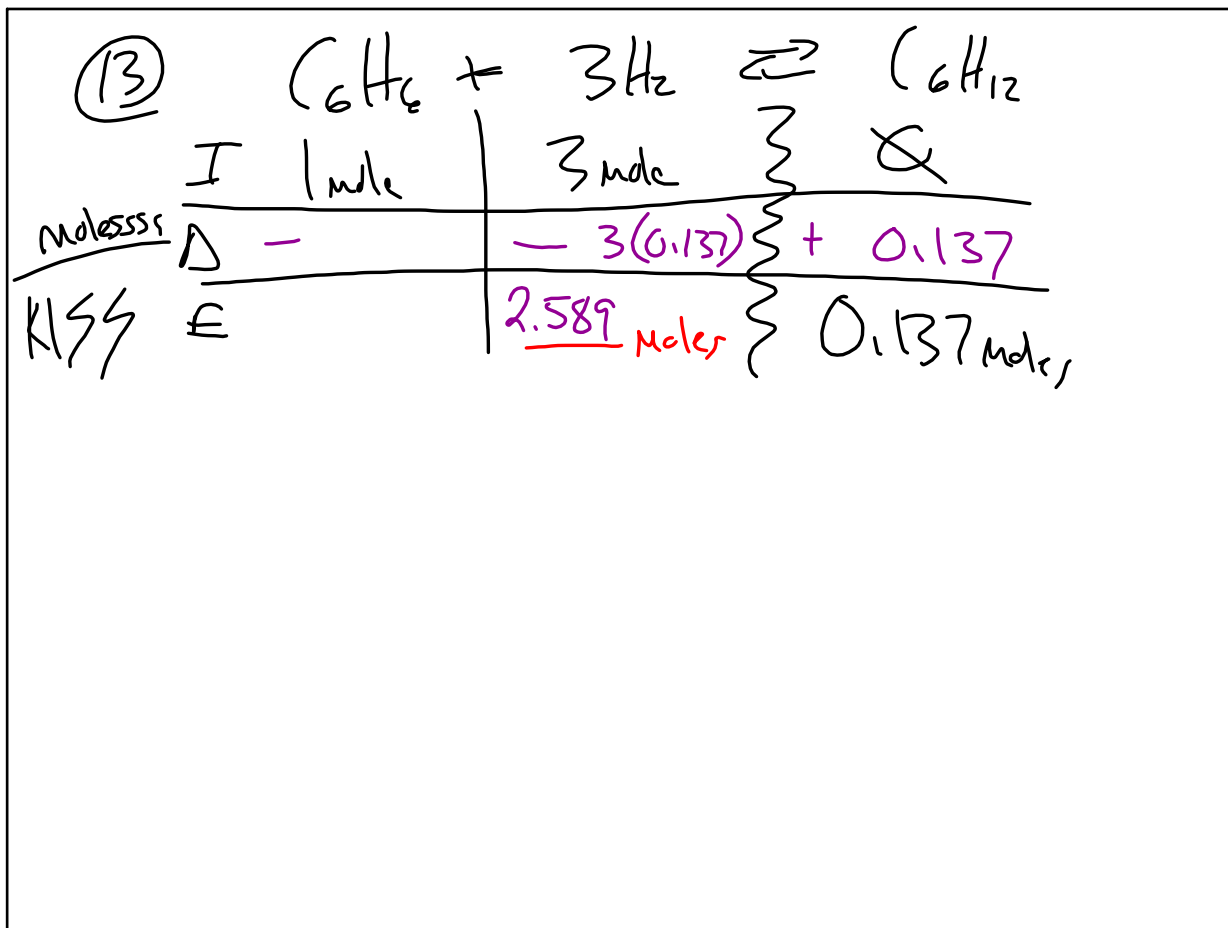
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$$\textcircled{9} \quad \ln \frac{k_1}{k_2} = \frac{E_a}{R} \left( \frac{1}{T_2} - \frac{1}{T_1} \right)$$

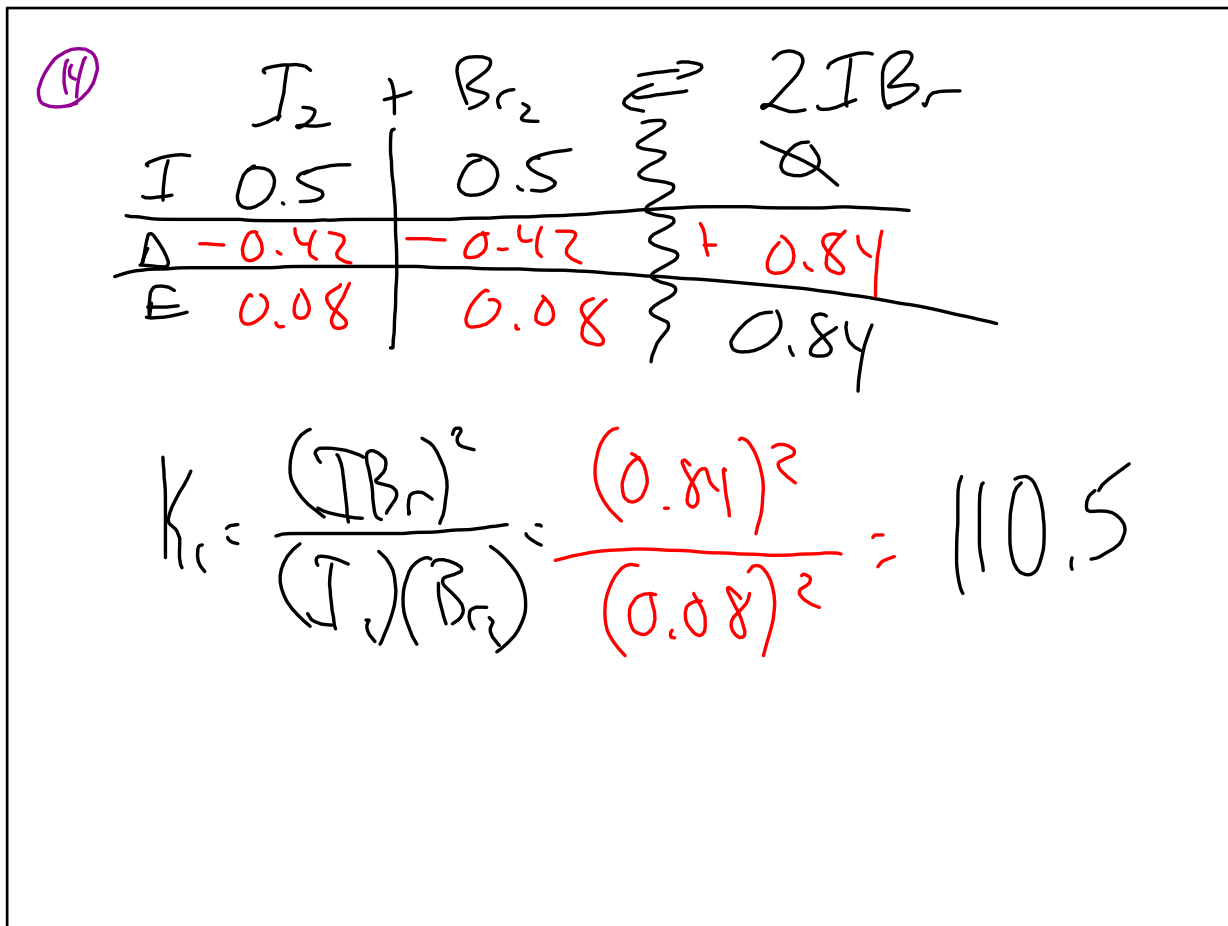
$$\ln \frac{3}{1} = \frac{E_a}{8.314 \times 10^{-3}} \left( \frac{1}{293} - \frac{1}{313} \right)$$

$$\left. \begin{array}{l} k_1 = 3 \quad T_1 = 40^\circ\text{C} \\ k_2 = 1 \quad T_2 = 20^\circ\text{C} \end{array} \right\}$$

Feb 13-8:57 AM



Feb 13-9:08 AM



Feb 13-9:12 AM



↑  
exo



decr. Heat

Want ↑ P, must ↓ V

Feb 13-9:15 AM